

## A NEW MINOR BUTENOLIDE FROM *Machilus odoratissima*

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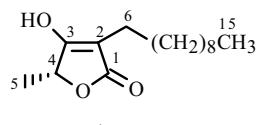
*A new butenolide, designated odoratinolide (**1**), was isolated from the bark of the Vietnamese medicinal plant Machilus odoratissima. Its structure was determined by spectroscopic analyses.*

**Keywords:** *Machilus odoratissima*, Lauraceae, butenolide.

In our previous papers [1, 2] the chemical profile of the *n*-hexane-soluble fraction of the MeOH extract of the bark of *Machilus odoratissima* Nees (Lauraceae) was found. Gradient chromatographic separation of this soluble fraction on silica gel gave mono- and sesquiterpenoids,  $\beta$ -sitosterol and stigmasterol [1], and lignans and neolignans [1, 2] in the order of increasing polarity. In the framework of our continuing study of the  $\text{CH}_2\text{Cl}_2$ -soluble fraction of the same MeOH extract, a new minor 3-hydroxybutenolide **1** was isolated. This paper discussed the isolation and structure elucidation of this compound.

Extraction and liquid-liquid fractionation of the MeOH extract of the dried bark of *M. odoratissima* gave the *n*-hexane,  $\text{CH}_2\text{Cl}_2$ , EtOAc, and 1-BuOH-soluble fractions. A procedure was established for the isolation of the minor butenolide **1**, including successive gradient column chromatography (CC) on silica gel and ODS (octadecyl silica gel) and ODS HPLC purification.

Compound **1** was obtained as an amorphous powder. The molecular formula of **1** was determined to be  $\text{C}_{15}\text{H}_{26}\text{O}_3$  by positive-ion HR-FAB-MS  $m/z$ : 255.1960 [ $\text{M} + \text{H}$ ]<sup>+</sup>. The IR spectrum of **1** showed absorption bands of hydroxyl groups ( $3382 \text{ cm}^{-1}$ ) and a double bond ( $1643 \text{ cm}^{-1}$ ). The <sup>1</sup>H NMR spectrum of **1** established the presence of a long alkyl chain [ $\delta$  0.79 (3H, t,  $J = 6.8 \text{ Hz}$ ), 1.17 (14H, br.s), 1.34 (2H, br.s), and 1.98 (2H, t,  $J = 7.8 \text{ Hz}$ )] and a secondary methyl group [ $\delta$  1.29 (3H, d,  $J = 6.6 \text{ Hz}$ ,  $\text{H}_3-5$ )] which was bonded to an isolated oxymethine [ $\delta$  4.39 (1H, q,  $J = 6.6 \text{ Hz}$ )]. The methylene group at  $\delta$  1.98 (2H, t) was clearly attached to a double bond. Analysis of the <sup>13</sup>C NMR spectrum of **1** showed the signals of a lactone ring ( $\delta$  178.9), a double bond ( $\delta$  131.9 and 147.2), and an oxymethine group ( $\delta$  75.7). The other carbon 13 signals were attributed to two methyl groups and the aliphatic methylenes of the long alkyl chain. On the basis of the spectroscopic data, three double-bond equivalents calculated from the molecular formula of **1** can be accounted for by an  $\alpha,\beta$ -unsaturated  $\gamma$ -lactone. Comparison of the NMR data revealed the related structure of **1** to the synthetic (+)-(5S)-3-dodecyl-5-methylfuran-2(5H)-one [3] except for an additional hydroxyl group at C-3 leading to the existence of an isolated oxymethine group in **1**. The butenolide core skeleton of **1** was also supported by the structures of the butenolides isolated from *Hortonia* species [4]. The stereochemistry at C-4 was assigned to the *R*-configuration by comparison of its  $[\alpha]$  with those of similar compounds [4, 5]. Thus, the absolute structure of **1**, which was designated odoratinolide, was determined as shown.



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## EXPERIMENTAL

**General Experimental Procedures.** Optical rotation was measured on a JASCO P-1030 digital polarimeter. FT-IR spectrum was recorded on a Horiba FT-710 spectrophotometer.  $^1\text{H}$  (400 MHz) and  $^{13}\text{C}$  NMR (100 MHz) spectra were recorded on a JEOL JNM- $\alpha$  400 NMR spectrometer with tetramethylsilane as an internal standard. Positive-ion HR-FAB-MS spectra were measured on a JEOL SX-102 mass spectrometer with PEG-400 as a calibration matrix. HPLC was performed with a JASCO PU-1580 pump and an UV-2075 Plus detector (set at 210 nm) using YMC ODS analytical ( $150 \times 4.6$  mm i.d.) and preparative ( $150 \times 20$  mm i.d.) columns at the corresponding flow rates of 0.5 and 5 mL/min. TLC glass plates (Merck, silica gel 60 F<sub>254</sub>) were used for analysis. Silica gel 60 (0.063–0.200 mm, Merck, Germany), and reversed-phase ODS (YMC, Japan) were used for CC.

**Plant Material.** The bark of *M. odoratissima* (voucher specimen No. HCTN 2000-6) was collected and identified by Dr. Nguyen Hoanh Coi (Military Center for Drug Control and Research, Hanoi, Vietnam) in June 2000 in Thai Nguyen Province, Northern Vietnam.

**Extraction and Isolation.** The air-dried bark of *M. odoratissima* (2.0 kg) was powdered and then extracted three times (each time for 3 days) with MeOH at room temperature. The MeOH extract was partitioned between  $\text{H}_2\text{O}$  and *n*-hexane,  $\text{CH}_2\text{Cl}_2$ , EtOAc, and 1-BuOH, successively, to afford the corresponding soluble fractions [2]. The  $\text{CH}_2\text{Cl}_2$ -soluble fraction (17.8 g) was chromatographed on a gradient silica gel column using  $\text{CHCl}_3$ –MeOH, 15:1, 10:1, 6:1, and 3:1 as solvent systems to afford four main fractions on the basis of their TLC pattern. Fraction 1 (1.8 g) was subjected to gradient column chromatography on ODS eluting with MeOH– $\text{H}_2\text{O}$ , 3:2, 3:1, and 4:1, and subfraction 1 was purified by using preparative ODS HPLC (MeOH– $\text{H}_2\text{O}$ , 3:1) to yield **1** (1.6 mg).

**Odoratinolide (1).** White amorphous powder,  $[\alpha]_D^{25} -1.75^\circ$  (*c* 0.16, MeOH). IR (film,  $\nu_{\max}$ ,  $\text{cm}^{-1}$ ): 3382, 1707, 1643, 1566, 1454, 1261, 1076. Positive-ion HR-FAB-MS *m/z*: 255.1960 [ $\text{M} + \text{H}]^+$  (calcd for  $\text{C}_{15}\text{H}_{27}\text{O}_3$ : 255.1961).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ,  $\delta$ , ppm, J/Hz): 0.79 (3H, t, *J* = 6.8, 3H-15), 1.17 (14H, br.s, 2H-8 – 2H-14), 1.29 (3H, d, *J* = 6.6, 3H-5), 1.34 (2H, br.s, 2H-7), 1.98 (2H, t, *J* = 7.8, 2H-6), 3.28 (1H, br.s, 3-OH), 4.39 (1H, q, *J* = 6.6, H-4).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ,  $\delta$ , ppm): 13.6, 17.5, 20.9, 22.4, 28.4, 28.9, 29.3, 29.4, 29.5 (C-5, C-7, C-8, C-9, C-10, C-11, C-12, C-13, C-14, C-15), 31.6 (C-6), 75.7 (C-4), 131.9 (C-2), 147.2 (C-3), 178.9 (C-1).

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## REFERENCES

1. M. G. Phan, T. V. H. Do, and T. S. Phan, *Proceedings of the Fourth Vietnam Conference on Organic Chemistry*, Hanoi, Vietnam, 2007, p. 334.
2. M. G. Phan, T. S. Phan, K. Matsunami, and H. Otsuka, *Chem. Pharm. Bull.*, **54**, 380 (2006).
3. A. M. E. Richecoeur and J. B. Sweeney, *Tetrahedron*, **56**, 389 (2000).
4. R. Ratnayake, V. Karunaratne, B. M. R. Bandara, V. Kumar, J. K. MacLeod, and P. Simmonds, *J. Nat. Prod.*, **64**, 376 (2001).
5. T. Momose, N. Toyooka, M. Nishio, H. Shinoda, H. Fujii, and H. Yanagino, *Heterocycles*, **51**, 1321 (1999).